

# Transfer hydrogenation of activated ketones using novel chiral Ru(II)-*N*-arenesulfonyl-1,2-diphenylethylenediamine complexes

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**Abstract**—A series of  $\alpha$ -keto esters and  $\alpha,\alpha,\alpha$ -trifluoromethyl ketones were reduced in high yields and excellent enantioselectivities under Ru-catalysed transfer hydrogenation using novel chiral *N*-arenesulfonyl-1,2-diphenylethylenediamine ligands.

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Chiral  $\alpha$ -hydroxy esters are important pharmaceutical intermediates and can be obtained through bioreduction or catalytic hydrogenation of the corresponding ketones.<sup>1</sup> Few literature examples of their preparation involving asymmetric transfer hydrogenation have been described.<sup>2</sup>

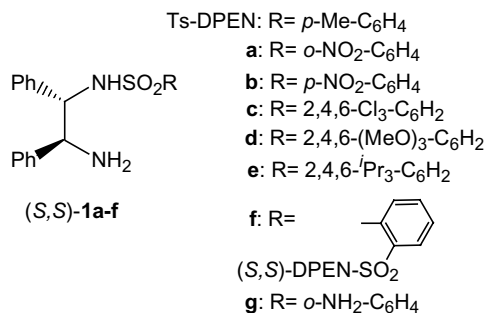
Asymmetric transfer hydrogenation using chiral Ru(II)-complexes with nitrogen-based chiral ligands proved to be a powerful method for the preparation of optically active  $\alpha$ -arylated and  $\alpha,\beta$ -acetylenic alcohols.<sup>3</sup> Such technology uses stable organic molecules such as 2-propanol or the formic acid–triethylamine azeotrope as hydrogen source. Amongst the various types of chiral nitrogen-based ligands, *N*-sulfonyl 1,2-diphenylethylenediamines (RSO<sub>2</sub>-DPENs) with R = Ar,<sup>4–8</sup> R<sub>F</sub>,<sup>9</sup> or R'<sub>2</sub>N<sup>10</sup> in combination with Ru(II)Cl( $\eta^6$ -arene) exhibit high enantiofacial discrimination ability. As proposed by Noyori and co-workers, the reduction takes place via a pericyclic mechanism involving an Ru hydride.<sup>11</sup> The RSO<sub>2</sub> functionality in the DPEN series is crucial in the reduction affecting the nucleophilicity of the neighbouring nitrogen and of the NH<sub>2</sub> in the Ru complex.

We disclose herein the preparation of a novel series of chiral ArSO<sub>2</sub>-DPEN ligands and their use in the Ru(II)-

catalysed transfer hydrogenation of a series of  $\alpha$ -keto esters and  $\alpha,\alpha,\alpha$ -trifluoromethyl ketones.

Chiral Ru(II)-complexes were prepared in situ at 80 °C in DMF<sup>10</sup> by reacting [RuCl<sub>2</sub>(*p*-cymene)]<sub>2</sub> with (1*S*,2*S*)-*N*-ArSO<sub>2</sub>-DPEN ligands **1a–f**, which were prepared in turn in DCM from the corresponding arenesulfonyl chloride and (1*S*,2*S*)-DPEN followed by chromatographic purification. These catalysts were used to reduce a variety of activated ketones<sup>12</sup> at room temperature using the formic acid–triethylamine azeotrope (Fig. 1, Table 1).

Methyl mandelate was obtained within 4 h in up to 96% ee using 0.5 mol % of the Ru-precursor and ligand **1e** or **1f** compared to 59% ee using Ts-DPEN. Similar results were obtained for ethyl mandelate. Whether in the presence of a Ru/ligand ratio of 1:1 or 2:1, complex of **1f** led to equivalent results. Ethyl 4'-fluoromandelate was

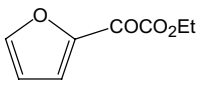


**Figure 1.** Novel series of (1*S*,2*S*)-*N*-arenesulfonyl-1,2-diphenylethylenediamine ligands based on Ts-DPEN.

**Keywords:** Asymmetric catalysis; Ketones; Ligands; Ruthenium; Transfer hydrogenation.

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**Table 1.** Asymmetric transfer hydrogenation of ketones<sup>a</sup>

Ketone	Ligand	Substrate/catalyst ratio	Time (h)	Conversion (%) <sup>b</sup>	Ee (%) <sup>b</sup>
C <sub>6</sub> H <sub>5</sub> COCO <sub>2</sub> Me	Ts-DPEN	200	4	100	59
	<b>1a</b>	200	4	90	85
	<b>1b</b>	200	4	100	71
	<b>1c</b>	200	4	100	93
	<b>1d</b>	200	4	100	76
	<b>1e</b> or <b>1f</b>	200	4	100	96
<i>p</i> -F-C <sub>6</sub> H <sub>5</sub> COCO <sub>2</sub> Et	Ts-DPEN	200	4	100	28
	<b>1b</b>	200	8	100	27
	<b>1c</b> or <b>1f</b>	200	4	100	77
	<b>1d</b>	200	8	100	70
	<b>1e</b>	200	4	100	84
<i>p</i> -MeO-C <sub>6</sub> H <sub>5</sub> COCO <sub>2</sub> Et	Ts-DPEN	200	24	34	73
	<b>1b</b>	200	24	44	75
	<b>1c</b>	200	24	58	95
	Ts-DPEN	40	36	100	82
	<b>1b</b> or <b>1e</b>	40	36	100	84
	<b>1c</b> or <b>1f</b>	40	36	100	94
	<b>1d</b>	40	36	100	87
C <sub>6</sub> H <sub>5</sub> (CH <sub>2</sub> ) <sub>2</sub> COCO <sub>2</sub> Et	Ts-DPEN	40	24	100	80
	<b>1b</b>	40	24	100	76
	<b>1c</b>	40	24	100	68
	<b>1e</b> or <b>1f</b>	40	24	100	74
CH <sub>3</sub> COCO <sub>2</sub> Et	Ts-DPEN	40	24	100	86
	<b>1b</b>	40	24	100	86
	<b>1c</b> or <b>1e</b>	40	24	100	78
	<b>1d</b>	40	24	85	74
	<b>1f</b>	40	24	100	83
(CH <sub>3</sub> ) <sub>2</sub> CHCOCO <sub>2</sub> Et <sup>c</sup>	Ts-DPEN	200	24	100	9
	<b>1c</b>	200	24	100	20
	<b>1f</b>	200	24	100	44
MeO <sub>2</sub> C(CH <sub>2</sub> ) <sub>2</sub> COCO <sub>2</sub> Me	Ts-DPEN	200	4	100	65
	<b>1b</b>	200	4	100	63
	<b>1c</b> or <b>1f</b>	200	4	100	60
	<b>1d</b>	200	4	50	65
	<b>1e</b>	200	4	100	58
C <sub>6</sub> H <sub>5</sub> COCF <sub>3</sub>	Ts-DPEN	100	24	100	42
	<b>1a</b>	100	24	100	42
	<b>1c</b> or <b>1f</b>	100	24	100	25
C <sub>6</sub> H <sub>5</sub> CH <sub>2</sub> COCF <sub>3</sub>	Ts-DPEN	100	1	100	95
	<b>1b</b> or <b>1c</b>	100	1	100	96
	<b>1d</b> or <b>1e</b>	100	1	100	97
	<b>1f</b>	100	1	100	98

<sup>a</sup> Reactions were carried out with HCO<sub>2</sub>H/Et<sub>3</sub>N 5:2 in DMF at 25 °C.

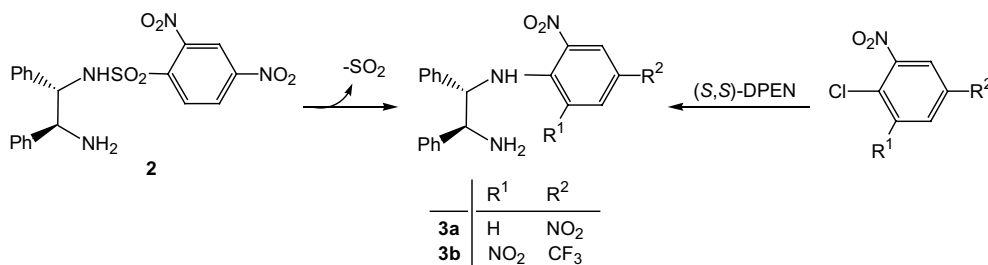
<sup>b</sup> Determined by GC analysis using Chirasil-DEX CB (25 m) unless otherwise stated.

<sup>c</sup> % Conversion and % ee determined by GC analysis using Chirasil-L-Val (25 m).

obtained in up to 84% ee using **1e** compared to 28% ee using Ts-DPEN. With an electron-donating group on the aromatic ring, the reduction was slower and ethyl 4'-methoxymandelate was obtained in 95% ee using **1c** compared to 73% ee using Ts-DPEN. The reduction of ethyl 2-furoylformate was sluggish requiring 2.5 mol % of Ru catalyst and led, after 36 h, to the corresponding  $\alpha$ -hydroxyester in up to 94% ee using **1c** or **1f** compared to 82% ee employing Ts-DPEN. An ACE inhibitor intermediate, ethyl 2-hydroxy-4-phenylbutanoate, was obtained in up to 76% ee compared to 80% ee using Ts-DPEN. The same enantioselectivity (86% ee) was

obtained for ethyl lactate using **1b** or Ts-DPEN. However, up to 44% ee was obtained in the reduction of ethyl ketovalinate using **1f** compared to 9% ee using Ts-DPEN. Moderate enantioselectivity (65% ee) was obtained in the reduction of dimethyl 2-oxoglutarate using **1d** or Ts-DPEN.

In the reduction of other activated ketones such as 2,2,2-trifluoroacetophenone, up to 42% ee was attained with **1a** or Ts-DPEN after 24 h. By contrast, the reduction of 1,1,1-trifluoro-3-phenyl-2-propanone took place within 1 h and led to 98% ee using **1f**.



Scheme 1. Synthesis of (1*S*,2*S*)-*N*-aryl-1,2-diphenylethylenediamine ligands.

Attempting to prepare *N*-(2,4-dinitrophenylsulfonyl)-DPEN ligand **2** following the standard procedure resulted, during purification, in the formation of *N*-(2,4-dinitrophenyl)-DPEN **3a** (Scheme 1).<sup>13</sup> No activity was observed for the corresponding Ru-complex under the standard conditions in the reduction of methyl benzoylformate as well as using the complex derived from *N*-(2,6-dinitro-4-trifluoromethylphenyl)-DPEN **3b**, which was prepared subsequently.

In addition, no reduction of methyl benzoylformate occurred using the *N*-(*o*-aminophenylsulfonyl)-DPEN ligand **1g** prepared by hydrogenation of the corresponding *N*-(*o*-nitrophenylsulfonyl)-DPEN ligand **1a**.

In summary, novel chiral *N*-arenesulfonyl-1,2-diphenylethylenediamine ligands were prepared and excellent enantioselectivities were obtained in the Ru(II)-catalysed transfer hydrogenation of  $\alpha$ -keto esters and  $\alpha,\alpha,\alpha$ -trifluoromethyl ketones. Enhanced enantioselectivity was reached by switching the tosyl group of Noyori's Ts-DPEN ligand with other arenesulfonyl groups and, in general, hindered ligands such as **1f** performed better.

### Acknowledgements

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- Proven by <sup>1</sup>H NMR spectroscopy before and after purification and also by its independent preparation using 1-chloro-2,4-dinitrobenzene; (1*S*,2*S*)-*N*-(2,4-dinitrophenyl sulfonyl)-1,2-diphenylethylenediamine (**2**): <sup>1</sup>H NMR  $\delta$  4.40 (d, 1H, CHNH<sub>2</sub>; *J* 3.2 Hz), 4.76 (d, 1H, CHNH; *J* 3.2 Hz), 6.98–7.42 (m, 10H, 2 $\times$ Ph), 7.64 (d, 1H, H-3; *J* 8.6 Hz), 8.08 (dd, 1H, H-5; *J* 8.6, 2.1 Hz), 8.45 (d, 1H, H-6; *J* 2.1 Hz).